# A 5-METHYLCOUMARIN GLUCOSIDE FROM ETHULIA CONYZOIDES\*

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Key Word Index-Ethulia conyzoides; Compositae; 5-methylcoumarin glycoside.

Ethulia conyzoides L. var. gracilis has been investigated before [1]. It is apparent that unusual 5-methylcoumarins are characteristic for this species. We have now examined the water-soluble fraction of an ethanolic extract of the aerial parts. Among other compounds, a crystalline glucoside, mp 150° (water), was isolated. The <sup>1</sup>H NMR data (see Table 1) of the tetraacetate clearly show that it is the coumarin 3. Therefore the natural compound is 2, the  $\beta$ -D-glucoside of the precursor 1 of all 5-methylcoumarins so far isolated from Ethulia conyzoides [1] and other composites [2–6]. The presence of a  $\beta$ -glucoside clearly follows from the observed coupling constants in the <sup>1</sup>H NMR spectrum of the tetraacetate, while in the spectrum of 2 the corresponding signals overlap. The signals of the aromatic protons are very similar to those of

Table 1. <sup>1</sup>H NMR data of 1 and 2 (270 MHz, TMS as internal standard)

	1 (CDCl <sub>3</sub> –DMSO)	<b>2</b> (CDCl <sub>3</sub> )
3-H	5.83 s	5.83
6-H	7.00 d(br)	7.05 d(br)
7-H	7.36 dd	7.41 dd
8-H	7.08 d(br)	7.20 d(br)
9-H	2.62 s	2.57 s
1'-H	5.37 d	5.34 d
2'-H )		5.43 dd
3'-H }	5.06 m	5.33 dd
4'-H		5.19 dd
5'-H	3.18 m	3.95 ddd
6 <sub>1</sub> '-H }	3.67 m	4.33 dd
6,′-H ∫		4.12 dd
OAc	_	2.13 s
		2.08 s
		2.07 s
		2.05 s

J (Hz): 6.7 = 7.8 = 8; 1'.2' = 7; 2'.3' = 9; 3'.4' = 10; 4'.5' = 9;  $5'.6_1' = 6.5$ ;  $5'.6_2' = 2.5$ .

the known coumarins [1,2]. Also the chemical shift of the 5-methyl group is in agreement with the proposed structure.

R = H R = Glc

 $3 R = Glc (OAc)_4$ 

$$\mathbf{A} = \mathbf{B} = \mathbf{A}\mathbf{c}\mathbf{O} + \mathbf{A}\mathbf$$

#### **EXPERIMENTAL**

<sup>1</sup>H NMR: 270 MHz; MS: 70 eV. Aerial parts (4.75 kg; collected near Alexandria on the banks of the Nile) were chopped and extracted with EtOH (room temp.). After evapn the residue was extracted with  $H_2O$  and the  $H_2O$ -soluble fraction was extracted several times with EtOAc yielding 11.3 g extract, which was separated by column chromatography (SiO<sub>2</sub>) with EtOAc-2% MeOH. 1 (60 mg) was obtained as colourless crystals, mp 150° (from  $H_2O$ ). IR  $\nu_{\text{max}}^{\text{nuiol}}$  cm<sup>-1</sup>: 3500–3200 (OH), 1680, 1615 and 1605 (coumarin); MS: M<sup>+</sup> m/e—; A 176.047 (61%) ( $C_{10}H_8O_3$ ); 134 (100) (176 – ketene).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{-117.5} \frac{578}{-119.6} \frac{546}{-137.5} \frac{436 \text{ nm}}{-254.6}$$

(c = 0.76, MeOH).

1 (20 mg) was acetylated with 1 ml Py and 0.2 ml Ac<sub>2</sub>O (24 hr, room temp.) yielding 20 mg 2, colourless crystals, mp 115–117° (CHCl<sub>3</sub>–Et<sub>2</sub>O). IR  $\nu_{\rm max}^{\rm CHCl_3}$  cm<sup>-1</sup>: 1760 and 1255 (OAc); 1717, 1620 and 1605 (coumarin); MS: M<sup>+</sup> m/e—; B 331 (1%); A 176 (12); 134 (19) (176 – ketene); 43 (100) (MeCO<sup>+</sup>); CI (isobutane): M<sup>+</sup> + 1501 (4%) (C<sub>24</sub>H<sub>26</sub>O<sub>12</sub>); B 331 (100); 271 (22) (331

<sup>\*</sup> Part 20 in the series "Naturally Occurring Coumarin Derivatives". For Part 19 see Rustaiyan, A., Nazarians, L. and Bohlmann. F. (1980) Phytochemistry 19, 2007.

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-HOAc); 1 A + 1 177 (23); 169 (41) (271 – ketene and HOAc); 109 (35) (169 – HOAc).

#### REFERENCES

- 1. Bohlmann, F. and Zdero, C. (1977) Phytochemistry 16, 1092.
- Balbea, S. I., Halim, A. F., Halaweish, F. T. and Bohlmann, F. (1979) Phytochemistry 18, 912.
- 3. Bohlmann, F. and Zdero, C. (1977) Phytochemistry 16, 1261.
- 4. Bohlmann, F. and Zdero, C. (1977) Chem. Ber. 110, 1755.
- 5. Bohlmann, F., Zdero, C. and Franke, H. (1973) Chem. Ber. 106,
- 6. Bohlmann, F. and Zdero, C. (1977) Phytochemistry 16, 239.

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## A XANTHONE-O-GLYCOSIDE FROM ASPLENIUM ADIANTUM-NIGRUM

### FILIPPO IMPERATO

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**Key Word Index**—Asplenium adiantum-nigrum; Aspleniaceae; 3.7.8-trihydroxyxanthone-1- $0-\beta$ -laminaribioside.

**Abstract**—A new xanthone-O-glycoside isolated from the fern Asplenium adiantum-nigrum has been identified as 3,7,8-trihydroxyxanthone-1-O- $\beta$ -laminaribioside by chemical and spectroscopic methods.

Recently a new hydroxycinnamic acid-sugar derivative, 1-caffeyllaminaribiose has been characterized from Asplenium adiantum-nigrum L. [1]. The present paper describes the identification of a new xanthone-O-glycoside, which has the same dissaccharide present, from the same fern.

The xanthone (colour reactions: orange-brown to yellow in UV + NH<sub>3</sub>) was isolated by prep. PC of an ethanolic extract of the fronds of Asplenium adiantum-nigrum. The UV spectral data:  $\lambda_{max}^{MeOH}$  240, 260, 315 and 364 nm; +NaOAc 260, 360 nm (increase in intensity);  $+ AlCl_3 245, 264, 349 (sh), 405 nm; + AlCl_3/HCl_2 243, 262,$ 340 (sh),  $405 \, \text{nm}$ ; + NaOAc/H<sub>3</sub>BO<sub>3</sub> 242, 258, 355 (sh), 418 nm, are consistent with a xanthone skeleton. The presence of a 3-hydroxyl group and an ortho-dihydroxyl group in the 7,8-position is indicated by the shifts with NaOAc, AlCl<sub>3</sub>, AlCl<sub>3</sub>/HCl and H<sub>3</sub>BO<sub>3</sub>/NaOAc [2]. Both total acid hydrolysis and treatment with  $\beta$ -glucosidase gave D-glucose and an aglycone identified as 1,3,7,8tetrahydroxyxanthone (norswertianin). This xanthone was first isolated in the free state from Swertia japonica [3] but has since been reported from Gentiana bavarica L. [4]. Norswertianin has been found as the 1-O-glucoside and 1-O-primeveroside in some Swertia and Gentiana species [5]. Controlled acid hydrolysis gave D-glucose and a disaccharide which was identical with an authentic sample of laminaribiose  $(3-O-\beta-glucosyl-D-glucose)$  synthesized according to Bächli and Percival [6]. Thus the isolated compound must be 3,7,8-trihydroxyxanthone-1- $O-\beta$ -laminaribioside, a new natural product. The structure of this substance was confirmed by methylation followed by acid hydrolysis to give 2,4,6-tri-O-methyl-D-glucose, 2,3,4,6-tetra-O-methyl-D-glucose and a partially methylated aglycone identified as 1-hydroxy-3,7,8-trimethoxyxanthone (decussatin) [7-9]. Xanthones have only been reported twice in ferns, in Asplenium montanum [10] and Athyrium mesosorum [11, 12]. However, the presence of a xanthone-O-glycoside in ferns is here reported for the first time. The absence of xanthones in Asplenium adiantum-nigrum L. collected in Astorias, Spain [10] was not confirmed in the present study. The chemical differences in plants from this locality may be due to phytogeographical factors. Since xanthones have been found previously only in Asplenium montanum among the several species of Asplenium surveyed [10], the isolation of a xanthone from A. adiantum-nigrum L. may suggest a relationship to A. montanum. Laminaribiose is here reported for the first time in association with xanthones. The other disaccharides found to date in xanthone-Oglycosides [5] are rutinose and primeverose.